Material Safety Data Sheet

MSDS

Product:

Prepared on:

Revised on:

Carbon dioxide Oiquefied

msds serial number carbon dioxide)
NTG-C02-02

September 25, 2007 February 1, 2017

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1. Chemical Product and Company Identification

Product name: Carbon dioxide (liquefied carbon dioxide)

MSDS serial number: NTG-C02-02

Supplier/Manufacture: Nippon Tansan Gas Co., Ltd.

Address: 3-12-15 Aoi, Adachi-ku, Tokyo, Japan

Division: Design Development Section, Technology Department

Phone: 03-3849-1571 Fax: 03-3880-6829

Emergency calls: 0282-27-5205

# 2. Composition and Information on Ingredients

Single or compound

Material: Single

Chemical name: Carbon dioxide
Content (vol%): 99.5 or more

Chemical formula: COz

Serial numbers of Law on the Examination and Regulation of Manufacture, etc., of Chemical

corresponding Substances: 1-169 (Japan)

official daily

gazettes:

CAS number: 124-38-9

Hazardous or toxic

components: Carbon dioxide

#### 3. Hazards Identification

Highest-level danger or toxicity:



Compressed gas (Contents under high pressure)

Health: Inhalation of high-purity of carbon dioxide may cause adverse effects on

the human body.

If liquefied carbon dioxide is released into the atmosphere, it turns into a low-temperature gas and snow-like dry ice, exposure to which may cause frostbite, and if such cold gas or dry ice enters the eyes, it may cause

blindness.

Note that the gas is asphyxiating even at low toxicity.

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Environmental

· Carbon dioxide is said to be a major cause of global warming, and measures

impact:

to reduce it are being considered domestically and internationally\_

Physical or

· Carbon dioxide is filled in containers for supply to customers.

chemical danger:

Because it is under high pressure, it is hazardous.

4. First Aid Measures

Inhalation: Immediately move the victim to fresh air. Loosen clothing and keep warm

> with a blanket or similar. If the victim is unconscious, loosen clothing, clear the respiratory tract, and conduct artificial respiration\_ Seek immediate

care by a physician.

If breathing is weak, give pure oxygen.

If not breathing, give artificial respiration.

If the person is lightly frostbitten, rub the affected area to warm it. Skin contact:

> If frostbite is serious, do not rub, but warm the affected area with tepid water, and wrap in gauze or similar. Seek immediate care by a physician.

Eye contact: If the victim has been exposed to a gas flow, wash the part with clear water.

Seek immediate care by a physician.

Protective

Ensure sufficient ventilation, and provide respirators and other equipment

measures before

starting first aid:

as needed.

5. Fire Fighting Measures

Use spray powder extinguisher and/or water. Use appropriate Extinguishing

media: extinguishing media for surrounding fire.

Specific hazard

necessary even though the substance is nonflammable:

or toxicity:

Move containers from fire area if it can be done without risk.

If it is not possible to move containers, turn water on them from windward.

For nearby fire, do as follows to prevent internal pressure from rising; this is

Keep away from leakage because of the risk of suffocation.

Protective

Using fireproof gloves, respirator, and other protective equipments,

measures

fight fires from windward and keep appropriate distance.

extinguishing a

fire:

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# 6. In Case of Leakage (Accidental release measures)

Respiration: Promptly evacuate personnel near leakage, keep others away, and ensure

sufficient ventilation.

Stay windward and dispose.

Protectors: If oxygen concentration is low, do not enter the area without protection.

Environmental Carbon dioxide is about 1.5 times heavier than air and tends to remain

Affects: in low places, resulting in high concentration.

High concentration may cause suffocation.

· Ventilate the room and let in fresh air. Discharge:

·Work in teams of two or more and wear protective equipment such as

respirators, protective gloves, and other proper guards.

# 7. Handling and Storage

Handling: Protection for carbon dioxide users

Suffocation

· Use in a place where ventilation is good.

## Handling of containers

Do not use containers roughly.

Do not use near fire.

Before using confirm the name of the gas by checking the mark or the other items on the containers.

Feed gas via a pressure regulator not directly.

Use only specialized pressure regulators.

Before connecting a pressure regulator, check the screw type.

Before using a gas container, check the pressure regulator, hose, pipes, joints ,etc.,for leakage.

Do not refill gas.

Do not modify or erase marks or other items on containers. Do not peel off labels on containers.

Do not use gas containers in electric circuits. Be careful not to cause arc strike particularly.

Do not touch dry ice with the bare hands because it has an extremely low temperature.

Avoid compressed gas discharge. Direct high-pressure gas may cause injury to the body.

Do not use gas containers for other purposes.

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Storage

Storage conditions: Storage conditions

Keep container away from fire and spark sources.

Do not store containers near electric lines or grounding.

Store containers at a dry place that is well drained and ventilated.

Keep containers away from corrosive atmosphere.

Keep containers away from direct sunlight at an ambient temperature of

0 to 40°C.

Do not expose containers to rough handling or falling.

Control oxygen concentration in storage area at 18 vol% or more.

Safe container

material: · Use only container for compressed gas.

8. Exposure Control and Personal Protection

Acceptable Japan Society for Occupational Health: (2000 version) 5,000 ppm (TWA)

concentration: ACGIH: 5,000 ppm (TLV-TWA) (1999 version)

300,000 ppm (TLV-STEL) (1999 version)

NIOSH: 5,000 ppm (TWA)

300,000 ppm (STEL)

OSHA: 5,000 ppm (TWA)

Note: ACGIH: American Conference of Governmental Industrial Hygienists

NIOSH: National Institute for Occupational Safety and Health

OSHA: Occupational Safety and Health Administration

TLV: Threshold Limit Value

TWA: Time Weighted Average Concentration

STEL: Short Term Exposure Limit

TLV-TWA: time-weighted average

Expressed as the time-weighted average concentration (TWA) during regular working time of 8 hours a day, 40 hours a week, and the health of most personnel is not adversely affected even when repeatedly exposed to such conditions.

TLV-STEL: short-term exposure limit

Indicates the limit of concentration that does not cause:

- 1) Intolerable stimulus,
- 2) Chronic or irreversible damage to the living body, or
- Increased risk of occurrence of injury, accidents, loss of self-control, or significantly lowered work efficiency caused by

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anesthetic action even after continuous short-term (15-minute) exposure to that environment provided that the day's average exposure does not exceed the TLV-TWA.

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Measures for

facilities: Before using containers in enclosed place, install a ventilator or similar.

Protectors

Protectors for

respiration: Air respirator, oxygen inhaler, gas mask

Protectors for Not needed

hands: When handling low-temperature containers ,wear leather gloves.

Protectors for

eyes: ·To protect eyes ,wear goggles.

Protectors for skin

and body: Not needed

# 9. Physical and Chemical Properties

Appearance

Physical state: Liquid

Color: Colorless
Odor: Odorless

Explosiveness: Nonflammable Molecular weight: 44.01 g/mol

Specific gravity: 1.5 (air= 1)

0.93(water = 1)

Temperature for

change in physical

state

Triple point: -56.6°C at 0.518 MPa

Critical

temperature: 31.10C

Critical pressure: 7.382 MPa

Boiling point:  $-78.5^{\circ}$ C

(Sublimation)

Solubility: 0.878 CO<sub>2</sub>/H<sub>2</sub>O (at 20°C, 1 atm.)

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# 10. Stability and Reactivity

Stability:

Reactivity:

No special reactivity

Hazardous or

harmful

degradation

products:

None

# 11. Toxicological Information

Acute toxicity:

Inhalation

The gas is low in inhalation toxicity but causes symptoms of oxygen deficiency if concentration is high.

0.04%	Normal air
0.5% (TLV)	Limit of long-term safety
1.5%	Tolerable for an extended time without affecting operability
	and basic physiology, but calcium and phosphorus
	metabolism may be affected in some cases.
2%	Respiration becomes deeper.
3%	Operability drops. Physiological changes appear in
	variations in blood pressure, heartbeat, and other factors.
4%	Respiration becomes much deeper. Higher breathing,
	slight gasping. Considerable degree of discomfort.
5%	Extreme difficulty in breathing; serious gasping intolerable
	for most people; some feeling of nauseum. Toxicosis occurs
	after 30 minute's exposure.
7 to 9%	Limit of tolerance, resulting in violent gasping. In about
	15 minutes, the subject looses consciousness.
10 to <b>11</b> %	Disabled regulation; unconsciousness in about 10 minutes.
15 to 20%	Much more serious symptoms are seen, but not
	lethal within an hour.
25 to 30%	Respiration weakens, blood pressure drops, resulting in
	coma, lost reflexes, and paralysis. Death occurs in some
	hours.

Local physical

effects on skin,

eyes, etc.: None Sensitization: None

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Chronic or

long-term toxicity: · None

# 12. Ecological Information

Global warming

potential (GWP): 1

\* Global Warming Potential (GWP)

Index detailing effects on global warming

# 13. Disposal Consideration

#### Disposal of carbon dioxide

Gradually release in open air, because it may cause simple suffocation, even though the toxicity is low.

Too quick gas release is dangerous because it may produce dry ice, causing frostbite.

# Disposal of containers

If gas remains in containers, drive into the container cap with appropriate application to release gas and dispose of containers as incombustible.

Do not dispose of containers without first checking that all gas has been released

For empty containers, check that containers are open and dispose of as incombustible waste.

# 14. Transport Information

International

regulations

UN number: IMDG UN2037 / IATA UN1013

UN classification: Class 2.2 (Compressed gas, Non-flammable gas, Non-toxic gas)

# 15. Regulatory Information

Legal information

(on Japan)

High Pressure Gas

Safety Law: Production, sales, storage, transportation, consumption, disposal

**Industrial Safety** 

and Health Law: Production, storage, consumption

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Food Sanitation

Law:

Production, sales

Law for Safety of

Containers:

Transportation

Port Regulation

Law:

Transportation

Civil Aeronautics

Law:

Transportation

Road Trucking

Vehicle Law:

Transportation

Road Traffic Law:

Transportation

#### 16. Other Information

Scope:

This Material Safety Data Sheet applies to liquefied carbon dioxide.

Article 2 of the High Pressure Gas Safety Law (Japan) , Carbon dioxide

is specified as a "High Pressure Gas".

Data sheet:

This Material Safety Data Sheet (MSDS) is prepared based on the latest materials and data. It may be subject to change when new material

and/or data are obtained.

The MSDS states precautions assuming that the product is used under normal conditions. Uses under special conditions should take these conditions into account to ensure safety. While the MSDS has been prepared as comprehensively as possible, we cannot guarantee its applicability or effectiveness under all possible conditions or

applications.

History of revision:

Prepared on: September 25, 2007

Revised on: March 1, 2010

Bibliography:

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Data Book on Safe Handling of Gas jointly edited by Nippon Sanso Co., Ltd. and Matheson Gas Products, published by Maruzen (1988) Profiles of 100 Hazardous and Harmful Substances, edited by Kikuo

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